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## Calculation Refresher

Robert H. Hoy, Pharm.D., BCPS  
Clinical Associate Manager  
Pharmacy Department  
Decatur Memorial Hospital  
Decatur, IL

Speaker has nothing to disclose.

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## Ratio and Proportion Method

- A ratio is a relationship. 5mg / ml
- A proportion is an equation. 1:3 = 3:x
- Can be used to find an unknown value:
  - it has four terms, one of which is unknown (x)
  - units for each term need to be correct
  - cross multiplying is used to solve for “x”
- Word problems will be converted into a proportion (equation with four terms).

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1	=	3
3	=	x

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## To solve for x:

- $1 / 3 = 3 / x$  We are trying to determine the value of x.
- We know the answer, because it is an easy example.
- How can we use the proportion calculation to get the answer?
- cross multiply:  $1 \times x = 3 \times 3$
- $1(x) = 3 \times 3$
- $x = 9$

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1	=	3
3	=	x

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## Word Problems

- In real life, we use calculations to solve “word problems”.
- Often we are asked to provide a numeric answer to a verbal question.
- “E.D. needs an IM injection of (some drug) 750mg. I am sending a label to the printer. Can you get started in the IV hood?”

## What is the Concentration of the Drug Product we will use?

- We convert the word problem into the proportion calculation
- The drug comes as a 500 mg/ml vial.
- We want a dose of 750mg.
- We will have to use more than one vial.
- Set up the proportion calculation.

$$\frac{500\text{mg}}{1\text{ml}} = \frac{750\text{mg}}{x\text{ml}}$$

$$\frac{500}{1} = \frac{750}{x}$$

## Cross Multiply

- $500 \times x = 750 \times 1$
- $500x = 750$
- Divide both sides by 500 to solve for x.
- $\frac{500x}{500} = \frac{750}{500}$
- $x = \frac{750}{500} \quad x = 1.5$
- Check the units.  $x$  is in ml, so  $x = 1.5$  ml.

## Let's Try Another Example

- "ICU wants a 30 mEq KCl rider. We will put it into 100 ml of normal saline. We will give it over 3 hours."
- Gather your ingredients:
  - 20 ml vial of concentrated (2 mEq/ml) KCl
  - 100 ml bag of normal saline (the "piggyback")
  - Normal saline is 0.9% sodium chloride in water.

## Word Problem Conversion

- Recognize that this is a proportion calculation.
- You have KCl in some concentration and you need to calculate how many ml to use.
- Your units are going to be mEq and ml.
- Calculations can use mcg, mg, gm, units, mmol, mEq, gram%, ml, liter, etc.

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$$\frac{2 \text{ mEq}}{1 \text{ ml}} = \frac{30 \text{ mEq}}{x \text{ ml}}$$

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- ### Solve for the amount of KCl
- We need to calculate how many ml of concentrated KCl solution is to be added to the 100ml bag of normal saline.
  - Check the units. This means to be sure that both sides of the equation use the same units.
  - The answer (x) is in ml.
  - Leave the units off to make the proportion calculation less cluttered.

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$$\frac{2}{1} = \frac{30}{x}$$

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- ### Solve the Proportion Calculation
- $2 \times x = 30 \times 1, \quad 2x = 30$
  - Divide both sides by 2.
  - $\frac{2x}{2} = \frac{30}{2}, \quad x = 15$
  - Answer is in ml (of concentrated KCl)
  - Finish the preparation. Put the 15ml of KCl into the 100ml bag of 0.9% NaCl.

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- ### Pharmacist Check
- Put a foil on the port used to inject the KCl solution.
  - Leave the syringe used to withdraw the KCl drawn back to 15ml.
  - Put syringe (without needle) next to the labeled IV bag.
  - Leave the used vial of concentrated KCl next to the IV bag.

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- ### Calculate the Volume and Infusion Rate of the KCl rider?
- 15 ml KCl + 100 ml 0.9% NaCl = 115ml
  - The order is to run the rider over 3 hours.
  - The IV pump is programmed in ml / hour.
  - Use a proportion calculation to calculate the IV pump rate.
  - Infuse the 115 ml over 3 hours.

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$$\frac{115\text{ml}}{3\text{ hours}} = \frac{X\text{ ml}}{1\text{ hour}}$$

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- ### IV Pump Rate?
- Check the units.
  - The answer is in ml.
  - This is the number that is “dialed” into the IV pump.
  - Leave out the units for a clearer picture.
  - Cross multiply:  $115 \times 1 = 3 \times x$ .
  - Divide both sides of the equation by 3
  - $x = 38$

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- ### Bonus Questions
- Why do we run the KCl rider (piggyback) at 10 mEq per hour (30 mEq in 3 hours)?
  - Answer: To avoid cardiac arrhythmias
  - Why do we dilute KCl riders instead of giving KCl in a more concentrated form?
  - Answer: To avoid vein irritation

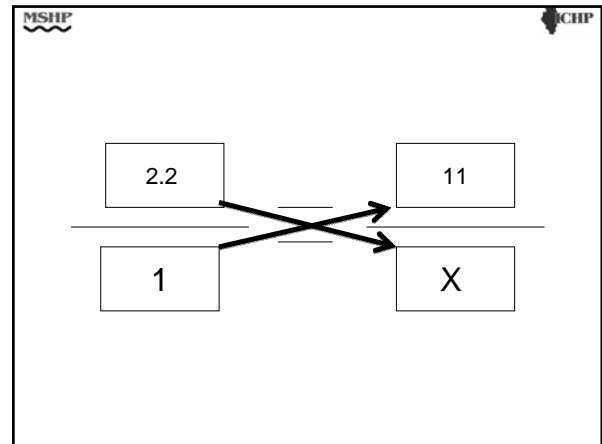
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- ### Pediatric Antibiotic Syringe
- “Pediatrics wants 80 mg / kg IV pip/tazo (Zosyn®) in a syringe.”
  - Zosyn® comes in a powder form in a 2.25 gram vial. We dilute it with sterile water.
  - The Zosyn® vial says to use 10 ml SWFI.
  - We put pediatric IV drugs in a syringe with 6 ml total fluid to infuse on a syringe pump
  - The patient weighs 11 lbs.

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- ### Proportion Calculations
- The patient’s mass in kg
  - The mg / kg dose of pip/tazo
  - The concentration of reconstituted pip/tazo
  - The dose of drug in ml of pip/tazo

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- ### The Patient’s Mass in kg
- There are 2.2 pounds per kg.
  - The patient weighs 11 pounds.
  - What units are in the equation?
  - What are the units in the answer?

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$$\frac{2.2 \text{ pounds}}{1 \text{ kg}} = \frac{11 \text{ pounds}}{x \text{ kg}}$$



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### Cross Multiply

- $2.2 X x = 1 X 11$
- $2.2x = 11$
- Simplify the side of the equation with the x by dividing BOTH sides by the number next to the x (2.2)
- $\frac{2.2x}{2.2} = \frac{11}{2.2}$
- $x = 5$ , So the patient weighs 5 kg.

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### The mg / kg Dose

- The dose is 80 mg / kg.
- The patient's mass is 5 kg.
- We need the dose in mg.
- Set up the proportion calculation.
- Check the units.
- The answer is in mg.

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$$\frac{80 \text{mg}}{1 \text{ kg}} = \frac{x \text{ mg}}{5 \text{ kg}}$$

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### Calculate the dose in mg

- Cross multiply
- $80 X 5 = 1 X x$
- $400 = x$
- The dose is 400mg.

## Units of pip/tazo per vial

- The label says 2.25 grams per vial.
- We need to calculate in mg, not grams
- There are 1000 mg/gm. Multiply 2.25 x 1000 to convert gm to mg.
- So, another way to state the amount of pip/tazo is 2,250 mg per vial.
- We reconstitute the vial with 10 ml of SWFI. Calculate mg/ml pip/tazo.

$$\frac{2,250\text{mg}}{10\text{ ml}} = \frac{x\text{ mg}}{\text{ml}}$$

## Managing the Equation

- Make sure the units are correct.
  - The units in the numerators (top) should be the same.
  - The units in the denominators (bottom) should be the same.
- Make a mental note of the units for x. This will always be part of the answer.
- Now leave out the units for calculations.

$$\frac{2,250}{10} = \frac{x}{1}$$

## Reconstituted pip/tazo

- $2,250 \times 1 = 10 \times x$
- $2,250 = 10x$
- Simplify by dividing both sides by 10.
- $\frac{2,250}{10} = \frac{10x}{10}$
- $225 = x$                        $x = 225\text{ mg}$
- The proper units for concentration are 225mg/ml.

## Dose of pip/tazo in ml

- The concentration of the reconstituted pip/tazo is 225 mg per ml.
- The dose is 400 mg.
- Calculate the dose in ml.

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$$\frac{225\text{mg}}{\text{ml}} = \frac{400\text{mg}}{x \text{ ml}}$$

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### Now Cross-multiply

- Check the units
- Answer is going to be in ml.
- Leave off the units for clarity

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$$\frac{225}{1} = \frac{400}{x}$$

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### Dose of pip/tazo in ml

- $225 X x = 1 X 400$
- $225x = 400$
- Divide both sides by 225 to isolate the x by itself.
- $\frac{225x}{225} = \frac{400}{225}$
- $X = 1.8$ ; Answer is in ml.

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### Dilute the Dose into 6 ml

- Our pediatric IV syringes are compounded with a total volume of 6 ml.
- Subtract the volume of pip/tazo from 6 ml.
- This is the volume of SWFI or NS or D5W to add to the syringe with the pip/tazo.
- $6 \text{ ml} - 1.8 \text{ ml} = 4.2 \text{ ml}$
- Add 4.2 ml to make a total volume of 6 ml.

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### Review

- Translate the word problem into a proportional calculation.
- Set up the equation.
- Make sure the units will cancel.
- Cross-multiply (without the units).
- Divide to eliminate the term with the x.

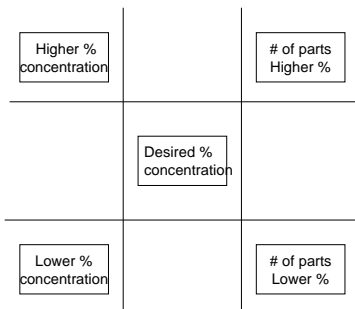
## PART 2

### A second technique - alligation

- Used when there are two liquids of different concentrations to be mixed to make a liquid with a concentration between the two.
- The concentration is in percent (%).
- The amount of each liquid is called a part.
- Add the parts to total the number of parts.

### Setting Up the Alligation

- Looks like a tic-tac-toe board
- 5 terms:
  - Three boxes for liquid concentrations in %
  - Two boxes for amount of liquid in parts
- Subtraction used to determine # of individual parts
- The units won't make Algebraic sense.

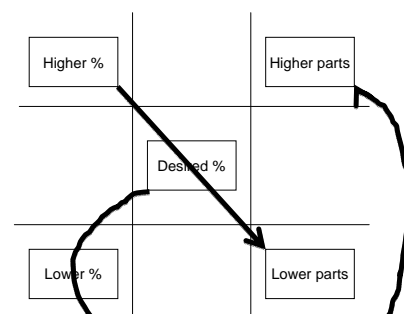


### Practice with %

- "one half" = 50%. So is 0.5,  $\frac{1}{2}$ , 1:2
- To convert a number to %, multiply x 100.
- $0.5 \times 100 = 50\%$
- $1:200 = 1/200 = 0.005$  Use your calculator.
- $0.005 \times 100 = 0.5\%$

### Order for Subtracting

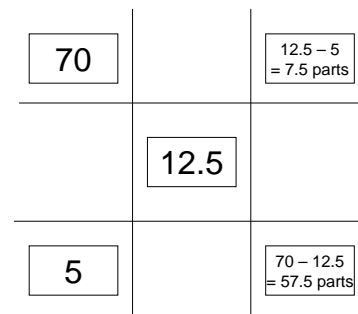
- Higher % - Desired % = # of parts of the Lower % liquid
- % - % = parts
- Desired % - Lower % = # of parts of the Higher % liquid





## Dextrose Alligation Problem

- “We need to make a liter of D12.5W.”
- We have D70W and D5W.
- D70W is 70% dextrose in water.
- 70% means 70 grams / 100 ml.
- Also called “gram %”
- D5W is 5% dextrose in water.
- 5% means 5 grams / 100 ml

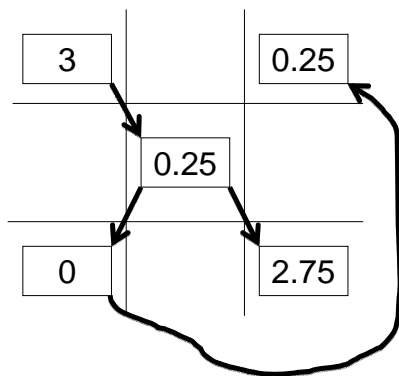


## Calculating the Volumes

- $57.5 \text{ parts} + 7.5 \text{ parts} = 65 \text{ parts total}$ .
- $57.5 / 65 = 0.88 \text{ (88\%)}$
- $7.5 / 65 = 0.12 \text{ (12\%)}$
- $88\% \text{ of } 1 \text{ liter} = 88\% \text{ of } 1,000 \text{ ml} = 880 \text{ ml}$
- $12\% \text{ of } 1 \text{ liter} = 12\% \text{ of } 1,000 \text{ ml} = 120 \text{ ml}$
- (Hint – Solve these using proportion calculations.)

## Try An Example Using 0%

- “E.D. wants half strength Dakin’s Solution”
- sodium hypochlorite topical
- Confusing units: For Dakin’s Solution, “full strength” =  $\frac{1}{2}\%$ . So, “half strength” =  $\frac{1}{4}\%$
- $\frac{1}{4}\% = 0.25\%$
- We have water for irrigation (0%) and 3% sodium hypochlorite (Clorox®)



## Calculating the Volume

- $2.75 + 0.25 = 3$
- $2.75 \text{ (parts of water)} + 0.25 \text{ (parts of bleach solution)} = 3 \text{ (parts of desired strength Dakins solution)}$
- $2.75 / 3 = 0.92 \text{ (92\%)} \text{ water}$ 
  - Made this way, Dakins is mostly water.
- $0.25 / 3 = 0.08 \text{ (8\%)} \text{ bleach}$ 
  - Clorox® is very concentrated. A poison.

## How to make 500 ml

- $2.75 / 3 = 0.92$  (Use a calculator)
- (Remember:  $0.92 = 92\%$ )
- $0.92 \times 500\text{ml} = 458$  ml water
- $0.25 / 3 = 0.08$  (Same thing as 8%)
- $0.08 \times 500\text{ml} = 42$  ml 3% sodium hypochlorite - Clorox®
- Carefully pour 42 ml bleach into 458 ml of water for irrigation.

## PART 3 - TPN Example

- Total Parenteral Nutrition (TPN)
- Using proportion calculations to make TPN
- TPN orders use a mixture of units.
- Some units refer to only part of the final mixture.
- The technician needs each ingredient converted into ml to make the TPN.

## TPN Orders

- 96 kg patient
- Total volume of TPN = 2,000 ml
- 20% lipid: 183 ml
- 15% amino acid: 0.5 gm / kg
- D70W: Use enough to make 75% of non-protein (lipid + carbohydrate) calories
- Electrolytes, vitamins, trace elements, etc.
  - Assume that these total 100 ml.

## Amino Acid Dose

- Dose = 0.5 gm / kg
- Patient weighs 96 kg
- Set up the proportion calculation
- $0.5 / 1 = x / 96$
- Check units; answer is in gm.
- We need to know ml, not gm!

$$\frac{0.5}{1} = \frac{x}{96}$$

$$\frac{0.5}{1} = \frac{x}{96}$$

### Grams of Amino Acid

- $0.5 \times 96 = 1 \times x$
- $0.5 \times 96 = x$
- $48 = x$
- Answer is in gm.
- We want ml.

### Volume of Amino Acid

- 15% amino acid = 15 gm / 100ml
- Dose = 48 gm
- Set up the proportion calculation
- $15 / 100 = 48 / x$
- Check units; answer is in ml

$$\frac{15}{100} = \frac{48}{x}$$

$$\frac{15}{100} = \frac{48}{x}$$

### ml of Amino Acid

- $15 \times x = 48 \times 100$
- $15x = 4,800$
- Get rid of 15 by dividing both sides by 15
- $\frac{15x}{15} = \frac{4,800}{15}$
- $x = 320$
- Answer is in ml.

### Carbohydrate Calories

- Dose = 75% of non-protein calories
- The sum of the lipid + carbohydrate calories is the non-protein calories
- We don't count the calories from the amino acids because we want them to be incorporated into new tissue (healing).
- Lipids provide 9 calories per gram.
- Carbohydrates provide 4 calories per gram

## How Many Lipid Calories?

- 20% lipid emulsion = 20 grams / 100ml
- Dose is 183 ml of lipid
- Set up the proportion calculation.
- $20 / 100 = x / 183$
- Check units; answer is in gm

$$\frac{20}{100} = \frac{x}{183}$$

$$\frac{20}{100} = \frac{x}{183}$$

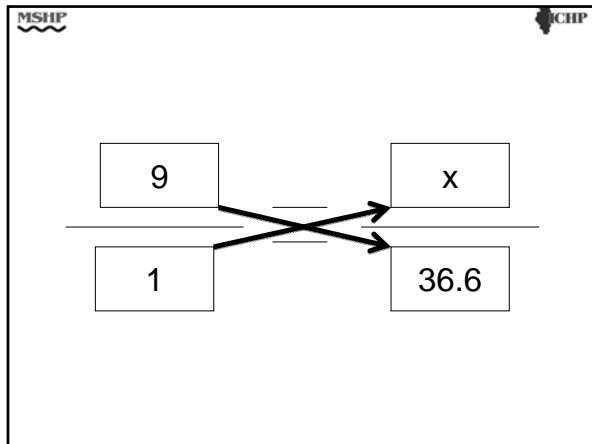
## gm of Lipid

- $20 \times 183 = 100 \times x$
- $3,660 = 100x$
- Get rid of 100 - divide both sides by 100.
- $\frac{3,660}{100} = \frac{100x}{100}$
- $36.6 = x$
- Answer is in gm

## Lipid Calories

- Dose = 36.6 gm of lipid
- There are 9 Calories per gm of lipid.
- Set up the proportion calculation.
- $9 / 1 = x / 36.6$
- Check units; answer is in Calories

$$\frac{9}{1} = \frac{x}{36.6}$$



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### Calories of Lipid

- $9 \times 36.6 = 1 \times x$
- $329.4 = x$
- Answer is in Calories (lipid calories)
- By knowing this, we can calculate the total number of (non-protein) Calories.
- The orders for carbohydrate Calories were associated with lipid Calories.

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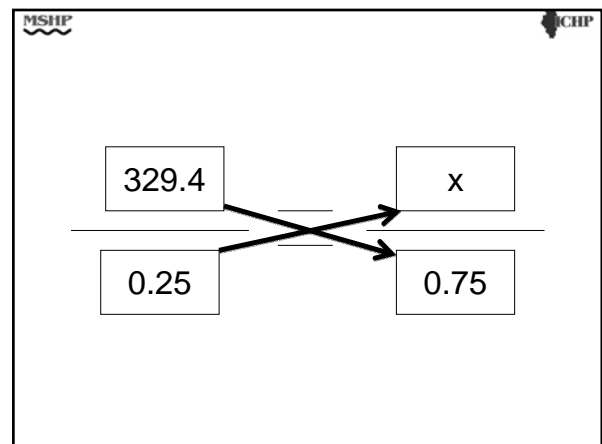
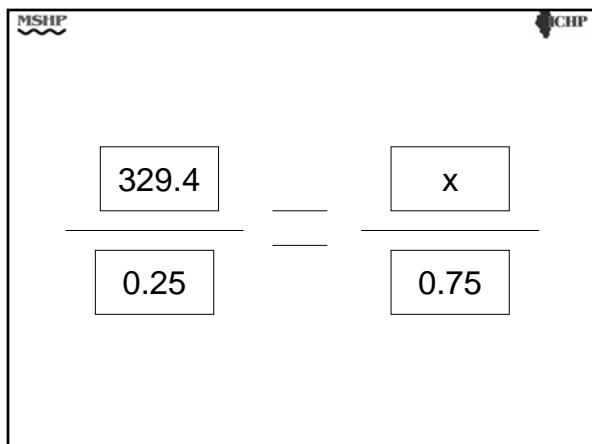
### Total Non-Protein Calories

- lipid Calories = 329.4 Calories
- carbohydrate Calories + lipid Calories = non-protein Calories (n-p C)
- carbohydrate Calories = 75% of n-p C.
- lipid Calories =  $100\% - 75\% = 25\%$  of n-p C
- $329.4$  lipid Calories = 25% of all n-p C.
- $25\% = 0.25$

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### n-p C from Carbohydrates

- $329.4$  lipid Calories = 0.25 total n-p C
- What are the n-p C from carbohydrate?
- Set up the proportion calculation
- $329.4 / 0.25 = x / 0.75$
- Check units; answer is in Calories
- Note: We do not have to calculate all of the n-p Calories; We only want the Carbs.



### n-p Calories from Carbohydrate

- $329.4 \times 0.75 = 0.25 \times x$
- $247 = 0.25 x$
- Get rid of the 0.25 by dividing both sides by 0.25
- $\frac{247}{0.25} = \frac{0.25 x}{0.25}$
- 988 Calories = n-p C from carbohydrate

### Grams of Dextrose

- 988 Calories from carbohydrates (sugar)
- 4 Calories per gram of dextrose (sugar)
- Set up the proportion calculation
- $4 / 1 = 988 / x$
- Check units; answer is in gm

$$\frac{4}{1} = \frac{988}{x}$$

$$\frac{4}{1} = \frac{988}{x}$$

Diagram illustrating the cross-multiplication process for the proportion  $\frac{4}{1} = \frac{988}{x}$ . Arrows show the 4 being multiplied by x and the 1 being multiplied by 988.

### Grams of Dextrose

- $4 \times x = 1 \times 988$
- $4x = 988$
- Get rid of the 4 by dividing both sides by 4
- $\frac{4x}{4} = \frac{988}{4}$
- $x = 247 \text{ gm} = \text{carbohydrate}$
- We are going to get it from D70W

### Volume of D70W

- Dose = 247 gm dextrose
- Product is D70W; 70% dextrose in water
- 70 gm dextrose / 100 ml water
- Set up the proportion calculation
- $70 / 100 = 247 / x$
- Check units; answer is in ml

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$$\frac{70}{100} = \frac{247}{x}$$

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$$\frac{70}{100} = \frac{247}{x}$$

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### ml of D70W

- $70 \times x = 100 \times 247$
- $70x = 24,700$
- Get rid of 70 by dividing both sides by 70
- $\frac{70x}{70} = \frac{24,700}{70}$
- $x = 353 \text{ ml}$

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### Sterile Water for Injection

- Total volume of TPN = 2,000 ml
- Lipid = 183 ml
- Amino Acids = 320 ml
- Dextrose = 352 ml
- Electrolytes, vitamins, trace elements, etc. = 100 ml
- $2,000 - (183 + 320 + 352 + 100) =$   
 $2,000 - 952 = 1,048 \text{ ml water}$

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### Helpful References

- Pharmacy Calculations, 2<sup>nd</sup> Edition  
– Mary F. Powers and Janet B. Wakelin  
– Morton Publishing Company, Englewood, CO  
– (Easy to understand. Separate section on Institutional Pharmacy, 250 pages)
- Reference Guide For Pharmaceutical Calculations – 3<sup>rd</sup> Edition  
– Manan Shroff, RXEXAM® (365 practice ?s)

**PRACTICE PROBLEMS FOR CALCULATIONS REFRESHER**

**121-000-09-016-L04-T**

**And Answer Key**

1. "Please make a one liter I.V. of D5NS with 10 mEq/l potassium chloride." The Pharmacy carries liters of D5NS and vials of concentrated KCl (2mEq/ml). **You would do the following:**
  - a. Add 20ml KCl to one liter of D5NS
  - b. Add 10ml KCl to one liter of D5NS
  - c. Add 5ml KCl to one liter of D5NS
  - d. Cannot be made with above ingredients
  
2. "A 20kg child in Pediatrics needs 50mg/kg ampicillin/sulbactam (Unasyn®) [a/s] every 6 hours." **What is the dose (mg)?**
  - a. 0.4
  - b. 2.5
  - c. 1,000
  - d. 1
  
3. A reconstituted vial of a/s contains 150mg/ml. **How many ml of a/s will you withdraw when preparing a dose of 750mg?**
  - a. 5ml
  - b. 7.5ml
  - c. 112.5 ml
  - d. 0.2ml
  
4. "Please make 250ml of hypertonic saline (3% sodium chloride in water)" The Pharmacy carries liters of NS (0.9% sodium chloride in water) and vials of concentrated sodium chloride (23.4% sodium chloride in water). **You would combine:**
  - a. 217.5 ml NS with 32.5 ml concentrated sodium chloride
  - b. 240 ml NS with 10 ml concentrated sodium chloride
  - c. 227 ml NS with 23 ml concentrated sodium chloride
  - d. 175 ml NS with 75 ml concentrated sodium chloride

The correct answers are:

1(c)

2(c)

3(a)

4(c)